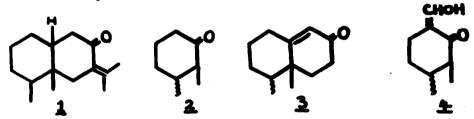
TOTAL SYNTHESIS OF (±)-FUKINONE

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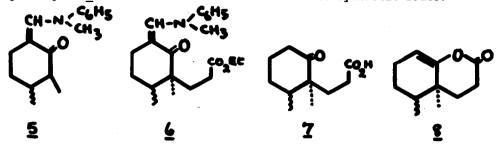
(+)-Fukinone is a bicyclic ketone isolated recently from flower stems of <u>Petasites japonicus</u> Maxim. A study of its chemical and spectral properties culminated in the structure and absolute stereochemistry depicted in $\underline{\mathbf{1}}$.¹ The ketone is a member of the ever-increasing family of sesquiterpenes derived from eremophilane.²



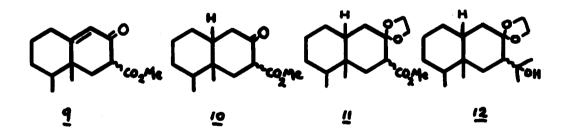
We describe here a stereospecific total synthesis of the racemic variety of fukinone (<u>1</u>). 2,3-Dimethylphenol was reduced catalytically³ to a mixture of epimeric 2,3-dimethylcyclohexanols, which on chromic acid oxidation³ afforded 2,3-dimethylcyclohexanone (<u>2</u>) (<u>cis and trans</u> isomers; approximately 6:1). Annelation of this mixture with methyl vinyl ketone⁴ yielded octalone <u>3</u>, in our hands in about 15% optimum yield, as a mixture of <u>cis</u> and <u>trans</u> forms (3:2).⁴ Attempts to separate these by fractional distillation through a spinning band column were only partially successful;⁵ a separation was ultimately achieved by fractional crystallization of the

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mixed semicarbazones. However, the low over-all yield of desired <u>cis</u> ketone corresponding to 3 necessitated a search for a better synthetic route.



Whilst our investigations to this end were in progress Piers and his collaborators⁶ described a synthesis of the <u>cis</u> form of <u>3</u>. We used their procedure with slight modification. The mixture of isomers <u>2</u> was condensed with ethyl formate⁶ to give the hydroxymethyleneketones <u>4</u>, treatment of which with <u>N</u>-methylaniline⁷ afforded the <u>N</u>-methylanilinomethyleneketones <u>5</u> (90% yield). Reaction of ethyl β -bromopropionate with <u>5</u> yielded the esters <u>6</u>, direct hydrolysis of which yielded the acids <u>7</u> (67%). Lactonization of the latter⁶ gave the epimeric lactones (89%), the <u>cis</u> isomer of which was obtained by fractional crystallization (76%). Exposure of this to methyl-lithium, followed by base treatment, yielded the pure <u>cis</u> ketone corresponding to 3 in 69% yield.⁶,8



Reaction of the foregoing <u>cis</u> octalone with dimethyl carbonate-sodium hydride furnished the β -ketoester 9 (81%), which, as anticipated, proved to

have enolic properties. Rapid catalytic hydrogenation of 9, in acid medium,⁹ generated the <u>cis</u>-2-decalone <u>10</u> (94%), with little evidence of formation of the <u>trans</u> isomer. This compound, which was also enolic in character, was converted into the ethylene ketal <u>11</u> in 85% yield in the usual manner. Reaction of the ketal with methyllithium gave the carbinol <u>12</u> (88%). Dehydration and hydrolysis of the latter to (t)-fukinone (<u>1</u>) was accomplished by boiling aqueous methanolic hydrochloric acid. The final product was purified by fractionation and column chromatography (57.5% yield; some retroaldol cleavage encountered), and finally through its crystalline semicarbazone,¹⁰ m.p. 200-201.⁰ It showed a single spot on thin layer chromatography, and its maximal u.v absorption, i.r. spectrum and n.m.r. spectrum were identical with the published data for the natural ketone.¹,¹¹

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- ⁵ We wish to thank Mr. B. J. Kane, Glidden Co., Organic Chemicals Division, Jacksonville, Florida, for valuable assistance with this fractionation.
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- ⁷ Cf. A. J. Birch and R. Robinson, J. Ch<u>em</u>. <u>Soc</u>., 501 (1944).
- ⁸ We wish to thank Dr. E. Piers, University of British Columbia, for establishing the identity of our <u>cis</u> ketone, after g.l.c. purification, with his product (ref. 6).
- ⁹ Cf. R. L. Augustine, J. Org. Chem., 23, 1853 (1958); 28, 152 (1963); "Catalytic Hydrogenation", Dekker, New York, 1965, p. 45; R. L. Augustine and A. D. Broom, J. Org. Chem., 25, 802 (1960).
- ¹⁰ Satisfactory analyses were obtained for all new compounds described herein.
- After the completion of our work alternative syntheses of racemic fukinone were described by J. A. Marshall and G. M. Cohen, <u>Tetrahedron Letters</u>, 3865 (1970), and by E. Piers and R. D. Smillie, J. <u>Org. Chem.</u>, <u>35</u>, 3997 (1970).